



## Tetrahedron Letters 40 (1999) 7231-7234

## On the Reduction Steps in the Mevalonate Independent 2-C-Methyl-D-erythritol 4-phosphate (MEP) Pathway for Isoprenoid Biosynthesis in the Bacterium Zymomonas mobilis

## Lionel Charon, Catherine Pale-Grosdemange and Michel Rohmer\*

Université Louis Pasteur/CNRS, Institut Le Bel, 4 rue Blaise Pascal, 67070 Strasbourg Cedex, France Received 15 July 1999; accepted 11 August 1999

Abstract. Zymomonas mobilis synthesizes triterpenoids of the hopane series via the methylerythritol phosphate (MEP) pathway. Owing to the peculiar metabolic pathways in this bacterium, [1-2H]glucose is the appropriate substrate for generating in vivo deuterium labeled NADP2H. After feeding of Z. mobilis with [1-2H]glucose, deuterium labeling was found on all carbon atoms derived from C-2 and C-4 of isopentenyl diphosphate. The deuterium at C-4 arose from the NADPH dependent reduction catalyzed by the reducto-isomerase converting 1-deoxy-D-xylulose 5-phosphate into 2-C-methyl-D-erythritol 4-phosphate. The deuterium at C-2 resulted from an additional reduction at a yet undetermined steps of the MEP biosynthetic pathway. © 1999 Published by Elsevier Science Ltd. All rights reserved.

Bacterial triterpenoids of the hopane series¹ played a key role in the discovery and the elucidation of the alternative mevalonate-independent route for isoprenoid biosynthesis (Scheme 1).² This metabolic route was found in bacteria, algae and plant plastids.³ Pyruvate 1 and glyceraldehyde 3-phosphate 2 (GAP) are the first precursors.²d A condensation catalyzed by a thiamin diphosphate dependent synthase yields 1-deoxy-D-xylulose 5-phosphate 3 (DXP).⁴ A reducto-isomerase catalyzes the next step: the rearrangement of the straight chain of DXP and the concomitant NADPH dependent reduction of the resulting aldehyde into 2-C-methyl-D-erythritol 4-phosphate 5 (MEP).⁵ Feeding experiments performed with ²H or ¹³C labeled deoxyxylulose showed that the free pentulose, or rather its 5-phosphate, was not only the precursor for isoprenoids,⁶ but also for thiamin diphosphate³ and pyridoxol phosphate.³ The formation of DXP does not accordingly represent the committed step in the non-mevalonate pathway. MEP, in contrast, seems devoid of any other role. Its biosynthesis might represent the key step of the mevalonate-independent route. Although free methylerythritol was incorporated into the isoprenoids of Escherichia coli,⁰ no data is available on other intermediates and reactions implied in the conversion of MEP into isopentenyl diphosphate 6 (IPP). Formally, the reaction sequence from MEP to IPP requires at least one phosphorylation, two reductions and the elimination of two molecules of water.

Hopanoids 10 and 11 (Scheme 1) are synthesized via the mevalonate-independent route in Zymomonas mobilis. <sup>2c</sup> This bacterium has no complete tricarboxylic acid cycle. <sup>10</sup> NADH and consequently NADPH, the cofactors involved in most biological reductions, are obtained either from the oxidation of glucose 6-phosphate into gluconate 6-phosphate lactone by the glucose 6-phosphate dehydrogenase or of GAP into 3-phosphoglyceric acid by the GAP dehydrogenase. In addition, glucose is metabolized via the Entner-Doudoroff pathway, and the resulting pyruvate is not converted into GAP because of the irreversibility of the reaction catalyzed by the enolase of this bacterium. Such a glucose catabolism implies that C-1 of glucose is completely lost as carbon dioxide by the decarboxylation of pyruvate and is not incorporated into the isoprenoids of Z. mobilis. <sup>2c</sup> When, however, [1-2H]glucose is utilized as sole carbon source, [4R-<sup>2</sup>H]- and [4S-<sup>2</sup>H]NADPH are formed by the action of the glucose 6-phosphate dehydrogenase, the GAP dehydrogenase and the two zinc or iron dependent alcohol dehydrogenases converting acetaldehyde into ethanol. <sup>11</sup> This makes

[1-2H]glucose a well suited substrate for studying the reduction steps of the MEP route in Z. mobilis. The bacterium was grown in the presence of [1-2H]glucose,  $^{12}$  and the hepta-acetates of the two major bacteriohopanetetrol derivatives 10 and 11 were isolated  $^{2c}$  and analyzed by  $^{1}$ H-NMR and  $^{13}$ C-NMR for deuterium induced  $\alpha$ -,  $\beta$ - and  $\gamma$ -shifts.  $^{13,14}$  Analysis of the labeling pattern was facilitated by the hopane skeleton. Indeed, it results from the cyclization of squalene 9 which is formed from two equivalent moieties derived from farnesyl diphosphate 8 and consequently symmetrical.

Scheme 1. Hopanoid biosynthesis in *Zymomonas mobilis via* the mevalonate-independent MEP pathway. Deuterium labeling introduced by (a) the deoxyxylulose reducto-isomerase, (b) an unknown enzyme involved in the formation of IPP, (O) the squalene synthase.

The NMR data of the hepta-acetates of hopanoids 10 and 11 was in accord with the presence of one deuterium on all carbon atoms derived from C-2 and C-4 of IPP with a ca. 50% isotopic abundance, as deduced from the integration of the signals from the C-23 and C-29 methyl groups in <sup>1</sup>H-NMR and the relative intensities of the <sup>13</sup>C signals presenting β-shifts. <sup>14</sup> Direct evidence for the presence of a deuterium was obtained for the C-23 and C-29 methyl groups. In <sup>1</sup>H-NMR, the non-labeled methyl signal was accompanied by an upfield shifted -CH<sub>2</sub>D signal, 14a and, in the 13C-NMR spectrum, the singlet, corresponding to the unlabeled methyl group, was accompanied by a triplet from the -CH<sub>2</sub>D group characterized by an upfield α-shift and a <sup>1</sup>J<sub>13C,D</sub> coupling constant. <sup>14c</sup> Similar α-shifts were observed for the signals of C-1, C-7 and C-19, confirming directly the presence of one deuterium on these carbon atoms. In the <sup>13</sup>C-NMR spectrum, nearly all carbon atoms corresponding to C-1 and C-3 of IPP were characterized by three signals in a 30:54:16 ratio: a singlet corresponding to no labeling on the two adjacent carbon atoms was accompanied by two additional singlets, corresponding to β-shifts induced either by the presence of a single deuterium on one of the adjacent carbon atoms derived from C-2 or C-4 of IPP, or by the simultaneous presence of a deuterium on each of these adjacent carbon atoms. For the C-10 and C-18 positions with three possible adjacent deuteriums, only two additional signals were found, the probability that the three adjacent positions were simultaneously labeled being too low. β-Shifts were expected for C-12 and C-17, because of the presence of deuterium at respectively C-13 and C-21, but were not observed. The presence of two an additional deuterium was also detected at C-11 and C-12. Indeed, the signals of C-9 and C-13 showed β-shifts induced by the deuterium atom introduced

during the reduction step catalyzed by the squalene synthase after the condensation of two farnesyl diphosphate moieties 8. As squalene 9 is symmetrical, this deuterium is present with the same probability at C-11 and C-12 on the hopane skeleton, and the intensity of the  $\beta$ -shifted signals was consequently lower for C-9 and C-13 than those of the signals described above. The numerous  $\gamma$ -shifts were in accord with the proposed localization of deuterium labeling.

All isoprenic units were identically labeled in the hopanoids. No difference was observed, whatever they were derived from IPP 6 or from dimethylallyl diphosphate 7 (DMAPP). The deuterium found on the carbon atoms derived from C-4 of IPP, corresponding to C-1 of MEP, was introduced by the DXP reducto-isomerase: this enzyme isomerizes DXP 3 into 2-C-methyl-D-erythrose 4-phosphate 4 which is concomitantly reduced by NADPH.<sup>5</sup> Presence of deuterium on the carbon atoms derived from C-4 of IPP confirmed that the catabolism of [1-2H]glucose allowed deuterium labeling of NADPH in Z. mobilis. The hydrogen atom found in isoprenic units on the carbon atoms derived from C-2 of IPP or DMAPP does not derive from DXP or from MEP. It is introduced at a later step, which has still to be identified, via a NADH or NADPH dependent reduction. These results fitted perfectly with those obtained after incubation of 1-deoxy-D-[2-\frac{13}{2}]C, 4-\frac{2}{4}|xylulose with a cell culture of Catharanthus roseus. The deuterium was lost in all isoprenic units from phytol and from lutein, which are synthesized via the MEP pathway. In contrast, the incubation of the [4-\frac{2}{4}|deoxyxylulose with E. coli showed two different labeling patterns for the isoprenic units from the prenyl side-chain of ubiquinone. Those derived from IPP were not labeled, whereas the terminal starter unit derived from DMAPP was characterized by a full retention of deuterium. No satisfactory explanation was proposed for this contradiction at the present stage of the elucidation of the metabolic route.

A similar deuterium incorporation was reported for the biosynthesis of sesquiterpenoids of the pentalenolactone series in *Streptomyces* UC5319, which are synthesized via the MEP pathway. <sup>17</sup> Distribution of the deuterium after feeding of [6,6-2H<sub>2</sub>]glucose showed a striking pattern: it was in contradiction with the MVA pathway, but fitted with the presence of reductions at the sites described above for the MEP pathway in *Z. mobilis. Streptomyces* utilizes glucose via glycolysis. Through the action of the aldolase and the triose phosphate isomerase, [6,6-2H<sub>2</sub>]glucose is converted into [1-2H]glucose, yielding finally pools of deuterium labeled NADH and NADPH. As observed in the case of the hopanoids of *Z. mobilis*, deuterium labeling was observed on all carbon atoms of the pentalenene derivatives derived from C-4 of IPP as well as on C-1 of pentalenolactone E corresponding to C-2 of IPP. All other carbon atoms derived from C-2 of IPP lost their deuterium labeling in the in pentalenolactone E or in other pentalenene derivatives by later oxidation steps of this biosynthetic pathway.

Experimental results indicated that IPP is not synthesized from DMAPP in Catharanthus roseus cell cultures, <sup>15</sup> and that IPP, and not DMAPP, might directly result in peppermint from the MEP pathway. <sup>18</sup> The IPP isomerase and the prenyl transferase present in E. coli the same enantioselectivity as the corresponding yeast enzymes, both eliminating the pro-R hydrogen of C-2 from IPP. <sup>19</sup> Assuming that the IPP isomerase and the prenyl transferases possess the same stereoselectivity in Z. mobilis and in E. coli, the retention of a deuterium in the isoprenic units of the Z. mobilis hopanoids after feeding of [1-2H]glucose suggested that the pro-R hydrogen of unknown origin at C-2 of IPP was removed, and consequently that the hydride labeled from [1-2H]glucose and introduced by reduction was in the pro-S position (Scheme 1).

Growth of Z. mobilis on [1-2H]glucose confirmed the presence in this bacterium of a reduction site on carbon atoms corresponding to C-4 of IPP: this was the signature of the DXP reducto-isomerase and indicated that deuterium labeled NADPH was synthesized from the glucose isotopomer. It also pointed out two other features. The hydrogen at C-4 of DXP or at C-3 of MEP is lost when these precursors are incorporated into the hopanoids, and the hydrogen found on all carbon atoms corresponding to C-2 of IPP is introduced by a NADH or NADPH dependent reduction. This hydrogen loss and this reduction occur at yet unidentified steps of the MEP pathway and represent a matter of further investigations. Labeling of the bacteriohopane side-chain yielded interesting clues on its biosynthesis and will be discussed elsewhere.

Acknowledgments. Our great thanks to Mr. Roland Graff and Mr. Jean-Daniel Sauer for the NMR measurements. This work was supported by a grant to M.R. from the "Institut Universitaire de France" and by a fellowship to L.C. from the "Ministère de l'Education Nationale et de la Recherche Scientifique".

## References and Notes

- 1. Rohmer, M., Pure Appl. Chem., 1993, 65, 1293-1298.
- (a) Flesch, G.; Rohmer, M., Eur. J. Biochem., 1988, 175, 405-411. (b) Rohmer, M.; Sutter, B.; Sahm, H., J. Chem. Soc. Chem. Commun., 1989, 1471-1472. (c) Rohmer, M.; Knani, M.; Simonin, P.; Sutter, B.; Sahm, H., Biochem. J., 1993. 295, 517-524. (d) Rohmer, M.; Seemann, M.; Horbach, S.; Bringer-Meyer, S.; Sahm, H., J. Am. Chem. Soc., 1996, 118. 2564-2566.
- 3. (a) Lichtenthaler, H.K.; Rohmer, M.; Schwender, J., Physiol. Plant., 1997, 101, 643-652. (b) Eisenreich, W.; Schwarz, M.; Cartayrade, A.; Arigoni, D.; Zenk, M.H.; Bacher, A., Chem. Biol., 1998, 5, R221-R233. (c) Rohmer, M., in Comprehensive Natural Products Chemistry. Isoprenoids including Carotenoids and Steroids, Cane, E.D., ed., Elsevier. Oxford, UK, 1999, vol.2, pp. 45-67. (d) Schwarz, M.; Arigoni, D., Ibid., pp. 367-400.
- Sprenger, G.A.; Schörken, U.; Wiegert, T.; Grolle, S.; de Graaf, A.A.; Taylor, S.V.; Begley, T.P.; Bringer-Meyer, S.; Sahm, H., Proc. Natl. Acad. Sci. USA, 1997, 94, 12857-12862. (b) Lois, L.-M.; Campos, N.; Rosa Putra, S.; Danielsen, K.; Rohmer, M.; Boronat, A., Proc. Natl. Acad. Sci. USA, 1998, 95, 2105-2110. (c) Lange, B.M.; Wildung, M.R., McCaskill, D.; Croteau, R., Proc. Natl. Acad. Sci. USA, 1998, 95, 2100-2104.
- 5. Takahashi, S.; Kuzuyama, T.; Watanabe, H.; Seto, H., Proc. Natl. Acad. Sci. USA, 1998, 95, 9879-9884.
- (a) Broers, S.T.J., Ph.D. Thesis Nb 10978, Eidgenössische Technische Hochschule, Zürich, Switzerland, 1994. (b) Arigoni.
  D.; Sagner, S.; Latzel, C.; Eisenreich, W.; Bacher, A.; Zenk, M.H., Proc. Natl. Acad. Sci. USA, 1997, 94, 10600-10605.
  (c) Rosa Putra, S.; Lois, L.-M.; Campos, N.; Boronat, A.; Rohmer, M., Tetrahedron Lett. 1998, 39, 23-26.
- (a) David, S.; Estramareix, B.; Fischer, J.-C.; Thérisod, M.C., J. Am. Chem. Soc., 1981, 103, 7341-7342.
  (b) Himmeldirk. K.; Sayer, B.G.; Spenser, I.D., J. Am. Chem. Soc., 1998, 120, 3581-3589.
- 8. Himmeldirk, K.; Kennedy, I.A.; Hill, R.E.; Sayer, B.G.; Spenser, I.A., J. Chem. Soc. Chem. Commun., 1996, 1187-1188.
- 9. Duvold, T.; Cali, P.; Bravo, J.-M.; Rohmer, M., Tetrahedron Lett., 1997, 38, 6181-6184.
- (a) Bringer-Meyer, S.; Sahm, H., FEMS Microbiol. Rev., 1988, 54, 131-142.
  (b) Sprenger, G.A., FEMS Microbiol. Lett., 1996, 145, 301-307.
- 11. Barrow, K.D.; Rogers, P.L.; Smith, G.M., Eur. J. Biochem., 1986, 157, 195-202.
- 12. Zymomonas mobilis (ZM6, DSM 3580, Deutsche Sammlung von Mikroorganismen und Zellkulturen, Braunschweig, FRG) was grown at 30°C for 20 h in anaerobic conditions on a synthetic medium (2 l, pH 5) containing [1-2H]glucose (10 g l<sup>-1</sup>, isotopic abundance 99%, Omicron Biochemicals Inc., South Bend, Indiana), KH<sub>2</sub>PO<sub>4</sub> (3.5 g l<sup>-1</sup>), MgSO<sub>4</sub>.7H<sub>2</sub>O (0.5 g l<sup>-1</sup>), NH<sub>4</sub>Cl (1.6 g l<sup>-1</sup>), (NH<sub>4</sub>)<sub>2</sub>Fe(SO<sub>4</sub>)<sub>2</sub>.6H<sub>2</sub>O (14 mg l<sup>-1</sup>), citric acid (0.2 g l<sup>-1</sup>), biotin (1 mg l<sup>-1</sup>), calcium panthotenate (1 mg l<sup>-1</sup>). The heptaacetates of hopanoids (10) (7.5 mg) and (11) (4.9 mg) were isolated from the lyophilized biomass (490 mg) as previously reported.<sup>2c</sup>
- 13. NMR spectra were recorded in CDCl<sub>3</sub> solution at 300 K on a Bruker ARX 500 spectrometer equipped with a Silicon Graphics station using CDCl<sub>3</sub> (δ=7.260 ppm) or <sup>13</sup>CDCl<sub>3</sub> (δ=77.030 ppm) as internal standard. <sup>13</sup>C-NMR spectra were measured as follows: 25° pulses (4 μsec); repetition time: 1.12 sec; spectra width: 26.3 kHz; data set: 32 kilo-words; zero filling to 34 kilo-words before Fourrier transformation; 1Hz experimental multiplication apodization; <sup>1</sup>H decoupling by WALTZ 16 during acquisition and relaxation.
- 14. Chemical shifts in the <sup>1</sup>H- and <sup>13</sup>C-NMR spectra and <sup>2</sup>H induced shifts in the <sup>13</sup>C-NMR spectrum were nearly identical for the hopane moiety in the spectra of the hepta-acetates of 10 and 11. For the sake of clarity, only the part of the spectrum concerning the triterpenic moiety of acetylated 10 is described. Data concerning the side chain will be reported elsewhere. <sup>1</sup>H-NMR (500 MHz):  $\delta$  (ppm) = 0.674 (3H, s, 18 $\alpha$ -CH<sub>3</sub>), 0.783 (3H, s, 4 $\beta$ -CH<sub>3</sub>), 0.806 (3H, s, 10 $\beta$ -CH<sub>3</sub>), 0.820 (1/2x2H, s. 4α-CH<sub>2</sub>D), 0.840 (1/2x3H, s, 4α-CH<sub>3</sub>), 0.868 (1/2x2H, d, J=6.0 Hz, 22-CH<sub>2</sub>D), 0.892 (1/2x3H, d, J=6.5 Hz, 22-CH<sub>3</sub>). 0.937 (6H, s, 8 $\beta$ - and 14 $\alpha$ -CH<sub>3</sub>). <sup>2</sup>H-NMR (76.7 MHz):  $\delta$  (ppm) = 0.89 (4 $\alpha$ - and 22-CH<sub>3</sub>), 1.20-1.85 (hopane skeleton). <sup>13</sup>C-NMR (126 MHz): δ (ppm)/<sup>2</sup>H induced shift (ppb) = C-1 (40.346/α-shift: -393, t, 20Hz), C-2 and C-6, (18.697/β-shifts: -105 and -198, biogenetically equivalent, both corresponding to C-1 of IPP), C-3 (42.086, broad signal/y-shifts: -14 and -32). C-4 (33.257/β-shifts: -77 and -160), C-5 (56.084, broad signal due to up to 5 γ-shifts, no detectable α-shift), C-7 (33.305/αshift: -364, t, J=20 Hz), C-8\* (41.818/β-shifts: -89 and -165), C-9 (50.461/β-shift: -38), C-10 (37.408/β-shifts: -93 and -179). C-11 (20.974, possible shifted signal concealed by those of the acetoxy methyl groups), C-12 (23.985), C-13 (49.337/β-shift: -45), C-14\* (41.700/β-shifts: -87 and -163), C-15 (33.708/possible α-shifted signal concealed by the signals of C-7, C-4 and C-23), C-16 (22.784/β-shifts: -83 and -188), C-17 (54.408, broadening of the peak, no measurable β-shift), C-18 (44.382/βshifts: -80 and -169), C-19 (41.569/ $\alpha$ -shift: -390, t, J=20 Hz), C-20 (27.603/ $\beta$ -shifts: -93 and -198), C-21 (45.946/n0 detectable  $\alpha$ - or  $\beta$ -shifts,  $\gamma$ -shift: -16), C-22 (36.150/ $\beta$ -shifts: -70 and 141) C-23 (33.411/ $\alpha$ -shift: -313,t, J=20 Hz), C-24 (21.593/r-shifts: -26 and -45), C-25 (15.910/r-shifts: -35 and -48), C-26° (16.615/r-shifts: -13 and -29), C-27° (16.536/r-shifts: -16 and -29), C-28 (15.875/ $\gamma$ -shifts: -29 and -42), C-29 (19.860/ $\alpha$ -shift: -300, t, J=19 Hz), C-30 (30.767/ $\gamma$ -shift: -16). Assignments of signals bearing the same superscript may be interchanged.
- Arigoni, D.; Eisenreich, W.; Latzel, C.; Sagner, S.; Radykewicz, T.; Zenk M.H.; Bacher, A., Proc. Natl. Acad. Sci. USA. 1999, 96, 1309-1314.
- 16. Giner, J.-L.; Jaun, B.; Arigoni, D., J. Chem. Soc., Chem. Commun., 1998, 1857-1858.
- (a) Cane, D.E.; Rossi, T.; Tillman, A.M.; Pachlatko, J.P., J. Am. Chem. Soc., 1981, 103, 1838-1843.
  (b) Schwarz. M.. Chimia, 1997,50, 280-281.
- 18. McCaskill, D.; Croteau, R., Tetrahedron Lett., 1999, 40, 653-656.
- 19. Leyes, A.E.; Baker, J.A.; Hahn, F.M.; Poulter, C.D., J. Chem. Soc., Chem. Commun., 1999, 717-718.